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# **B.SC DEGREE (C.B.C.S.S.) EXAMINATION, NOVEMBER 2014**

## **B.Sc. CHEMISTRY – FIRST SEMESTER-CORE COURSE**

#### CHE1MCDS- METHODOLOGY OF CHEMISTRY AS A DISCIPLINE OF SCIENCE

Time: Three Hours Maximum Marks: 60

#### **PART A**

# Answer all questions. Each question carries 1 mark.

- 1. Name one homologous series.
- 2. "Elements are made of tiny particles called atoms". This is a statement of \_\_\_\_ atomic theory.
- 3. What do you mean by IMRAD method in writing science?
- 4. Values close to +1 indicate \_\_\_\_ correlation.
- 5. -----is an indicator used in complexometric titration.
- 6. Give the full name of EDTA.
- 7. An accurate result is always precise. Say true or false?
- 8. To find the accuracy of a result ......is needed.

 $(8 \times 1 = 8)$ 

#### PART B

## Answer any six questions. Each question carries 2 marks.

- 9. State Avogadro's hypothesis.
- 10. What do you mean by falsification?
- 11. What is the difference between preparation and synthesis?
- 12. What is the significance of models in science?
- 13. Write a note on sampling in research.
- 14. Explain the concept of mole and molar volume.

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- 15. A solution is prepared by dissolving 2g NaOH in 250 ml water. Calculate the molarity of the solution.
- 16. Explain the terms mean and standard deviation.
- 17. Explain different types of errors in analysis.
- 18. What is the Q test for rejecting a suspected data?

 $(6 \times 2 = 12)$ 

#### PART C

# Answer any four questions. Each question carries 4 marks.

- 19. State and explain Charle's law?
- 20. "Correlation does not imply causality". Discuss.
- 21. How is a hypothesis developed?
- 22. What are errors in an experiment? How are they classified? What are the ways to reduce errors?
- 23. Explain the ways of expressing precision.
- 24. At a certain temperature the solubility product of AgCl is  $1\times10^{-10}$ . Calculate the solubility in g/litre at this temperature.

 $(4 \times 4 = 16)$ 

#### PART D

## Answer any two questions. Each question carries 12 marks.

- 25. Explain the role of chemistry as a central science.
- 26. Discuss the different methods of writing science.
- 27. Write a detailed account of various separation and purification techniques.
- 28. Briefly discuss about the tests of significance used in statistical analyses.

 $(2 \times 12 = 24)$