TB145670A	Reg. No:
	Name

B. Sc. DEGREE (C.B.C.S.S) EXAMINATION, OCTOBER 2016 SEMESTER V - PHYSICS PHY5TSP - THERMAL AND STATISTICAL PHYSICS

Time: Three Hours Maximum Marks: 60

PART A

Short answer questions

- I. Answer all questions. Each question carries 1 mark.
- 1. Explain the first law of thermodynamics.
- 2. What is an indicator diagram? State its importance.
- 3. What is the principle of unattainability of absolute zero?
- 4. Define entropy. Write down an expression for entropy in adiabatic process.
- 5. Give the concept of isothermal elasticity.
- 6. State the Wien's displacement law.
- 7. Give the concept degenerate energy levels.
- 8. Explain the need of quantum statistics.

(8x1 = 8)

PART B

Brief answer questions

- II. Answer any six questions. Each question carries 2 marks.
- 9. Derive an expression for the work done in during isothermal process.
- 10. What is a heat engine? What are its essential parts?
- 11. Derive an expression for the change in temperature of a metallic wire when it is stretched.
- 12. State and prove Carnot's theorem.
- 13. State and explain second law of thermodynamics.
- 14. Write down Claussius Clepeyron equation. Explain the terms.
- 15. Explain the different modes of heat transfer with examples.
- 16. What is meant by Micro and Macro states?
- 17. Explain the consequences of indistinguishability of particles in quantum statistics.
- 18. How we can apply Fermi-Dirac Statistics to electron gas.

(6x2 = 12)

PART C

Short essay questions

- III. Answer any four questions. Each question carries 4 marks.
- 19. A monatomic ideal gas of volume 1 litre and pressure 8 atmosphere undergoes adiabatic expansion until the pressure drops to 1atmosphere. What is the final volume and how much work is done? $1 \text{ atms} = 10^5 \text{Nm}^{-2}$.
- 20. A Carnot engine has an efficiency of 30% when the temperature of the sink is 27°C. What must be the change in temperature of the source to make its efficiency 50%.

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- 21. A Carnot reversible engine working between the temperatures 500 K and 300 K. Suppose 1000 joules of heat energy is drawn from the high temperature reservoir. Prove the Claussius Inequality = 0.
- 22. Calculate the probability that in tossing a coin 10 times, we get (i) all heads (ii) 5 heads & 5 tails.
- 23. The average kinetic energy of a gas molecule at a certain temperature is 6.21×10^{-21} joules. Find the temperature. Boltzmann's constant $K = 1.38 \times 10^{-23}$ joule K^{-1} .
- 24. Briefly compare the results of M.B, B.E and F.D statistics.

(4x4 = 16)

PART D

Essay type questions

- IV. Answer any two questions. Each question carries 12 marks.
- 25. Describe with necessary theory the working of a Petrol (Otto) engine. Calculate the efficiency of the engine.
- 26. Define thermodynamic potentials. Derive Maxwell's thermodynamical general equations connecting the thermodynamic quantities.
- 27. Define coefficient of thermal conductivity. Describe the Lee's method to find the coefficient of thermal conductivity of metals.
- 28. Derive Plank's law of black body radiation using Bose Einstein statistics.

(2x12 = 24)